

Carter problems 7.3, 7, 10, 15, 17

7.3) (a) $m = 0.4 \text{ kg}$, $C_p = 150 \text{ J/K}$, $T_0 = 100^\circ\text{C}$, $T_{\text{lake}} = 10^\circ\text{C}$

$$\left. \begin{aligned} \Delta S_{\text{block}} &= C_p \ln \frac{283}{373} = -414 \text{ J/K} \\ \Delta S_{\text{lake}} &= Q/T_{\text{lake}} = -\frac{C_p(T_{\text{lake}} - T_0)}{T_{\text{lake}}} = 47.7 \text{ J/K} \end{aligned} \right\} \Delta S_{\text{uni}} = 6.3 \text{ J/K}$$

(b) The block is unchanged $\rightarrow \Delta S_{\text{sys}} = 0$

$$\Delta S_{\text{lake}} = \Delta S_{\text{uni}} = \frac{mgh}{T_{\text{lake}}} = 1.38 \text{ J/K}$$

(c) Carter is wrong here. $\Delta S = C_p \left[\ln \frac{T_{\text{ave}}}{T_{100}} + \ln \frac{T_{\text{ave}}}{T_{10}} \right] = C_p \ln \frac{328^2}{283 \cdot 373}$

$$\Delta S_{\text{uni}} = 150 \cdot 0.019 = 2.85 \text{ J/K}$$

(d) $n = 1 \text{ kilomole}$, $V_0 \rightarrow 2V_0$

$$\Delta S_{\text{sys}} = nR \ln \frac{V_2}{V_0} = 5.76 \cdot 10^3 \text{ J/K (system)}, \Delta S_{\text{uni}} = 0 \text{ (reversible)}$$

(e) reversible + adiabatic \rightarrow isentropic, $\Delta S_{\text{uni}} = 0 = \Delta S_{\text{sys}}$

7.7) 50kg bag of sand at 25°C falls 10m to pavement.

(a) The change in gravitational potential energy shows up as dissipative work: $W_{\text{diss}} = -mgh = 4.9 \cdot 10^3 \text{ J}$ done by the sand
 $\rightarrow -4.9 \cdot 10^3 \text{ J}$ work done on the sand

(b) $Q = \Delta U + W$ so $\Delta U = 4.9 \cdot 10^3 \text{ J}$

(c) $T = \text{constant}$ so $\Delta S_{\text{sand}} = \frac{Q}{T} = \frac{\Delta U}{T} = 16.4 \text{ J/K}$

$$\Delta S_{\text{surf}} = 0 \text{ so } \Delta S_{\text{uni}} = 16.4 \text{ J/K}$$

7-10) (a) $T ds = du + P dv$. In a process at constant volume,

$$ds_v = \frac{1}{T} [du] = \frac{1}{T} \left[\left(\frac{\partial u}{\partial T} \right)_v dT + \left(\frac{\partial u}{\partial v} \right)_T dv \right] = \frac{C_v}{T} dT$$

Differentiating by P (dividing by dP_v) $\rightarrow \left(\frac{\partial s}{\partial P} \right)_v = \frac{C_v}{T} \left(\frac{\partial T}{\partial P} \right)_v$

Similarly, at constant pressure $ds_p = \frac{1}{T} [dh_p + v dp] = \frac{C_p}{T} dT + \left(\frac{\partial h}{\partial P} \right)_T dP$
 so $(\partial s / \partial P)_p = (C_p / T) (\partial T / \partial P)_p$

$$7-10)(b) \quad ds = \left(\frac{\partial s}{\partial P}\right)_v dP + \left(\frac{\partial s}{\partial v}\right)_P dv$$

$$= \frac{c_v}{T} \left(\frac{\partial T}{\partial P}\right)_v dP + \frac{c_p}{T} \left(\frac{\partial T}{\partial v}\right)_P dv$$

Since $\left(\frac{\partial T}{\partial P}\right)_v = \kappa/\beta$ and $\left(\frac{\partial T}{\partial v}\right)_P = \frac{1}{\beta v}$

$$\hookrightarrow T ds = c_v \frac{\kappa}{\beta} dP + \frac{c_p}{\beta v} dv$$

$$7-11) \quad \left(P + \frac{a}{v^2}\right)(v-b) = RT \quad \left(\frac{\partial P}{\partial T}\right)_v = \frac{R}{v-b}$$

$$T ds = c_v dT + T \left(\frac{\partial P}{\partial T}\right)_v dv \rightarrow ds = c_v \frac{dT}{T} + \frac{R}{v-b} dv$$

$$s(T, v) = s_0 + c_v \ln \frac{T}{T_0} + R \ln \frac{v-b}{v_0-b}$$

$$7-15)(a) \quad \eta \equiv \left(\frac{\partial T}{\partial v}\right)_o = - \left(\frac{\partial T}{\partial v}\right)_v \left(\frac{\partial v}{\partial T}\right)_T = -\frac{1}{c_v} \left[T \left(\frac{\partial P}{\partial T}\right)_v - P \right] = \frac{1}{c_v} \left[P - T \frac{\beta}{\kappa} \right]$$

$$(b) \quad \mu \equiv \left(\frac{\partial T}{\partial P}\right)_h = - \left(\frac{\partial T}{\partial h_P}\right) \left(\frac{\partial h}{\partial P}\right)_T = -\frac{1}{c_p} \left[v - T \left(\frac{\partial v}{\partial T}\right)_P \right] = \frac{v}{c_p} (T\beta - 1)$$

$$(c) \quad \text{van der Waals: } \left(\frac{\partial P}{\partial T}\right)_v = \frac{R}{v-b} = \frac{P + a/v^2}{T}, \quad \left(\frac{\partial v}{\partial T}\right)_P = \left(\frac{\partial T}{\partial v}\right)_P = \left[\frac{T}{v-b} + \frac{2a(v-b)}{Rv^3} \right]^{-1}$$

$$\eta = \frac{-a}{c_v v^2}, \quad \mu = -\frac{1}{c_p} \frac{RTb v^3 - 2av(v-b)^2}{RTv^3 - 2a(v-b)^2}$$

In the limit $a \rightarrow 0, b \rightarrow 0$ the van der Waals eqⁿ is $Pv = RT$ (ideal gas) so we have

$$\eta_{\text{ideal}} = 0, \quad \mu_{\text{ideal}} = 0$$

$$7-17) \quad v \approx v_0 \Rightarrow \beta, \kappa \text{ are small. } v - v_0 \approx 0 = (\beta \Delta T - \kappa \Delta P) v_0$$

$$(a) \quad ds = c_p \frac{dT}{T} - \left(\frac{\partial v}{\partial T}\right)_P dP = c_p \frac{dT}{T} - \left(\frac{\partial v}{\partial T}\right)_P \left[\left(\frac{\partial P}{\partial T}\right)_v dT + \left(\frac{\partial P}{\partial v}\right)_T dv \right] = c_p \frac{dT}{T} - \frac{\beta^2 v_0}{\kappa} dT$$

$$\Delta S = c_p \ln \left(\frac{T}{T_0}\right) - \frac{\beta^2 v_0}{\kappa} (T - T_0)$$

$$(b) \quad \text{Put in numbers: } 1^{\text{st}} \text{ term is } 12.79 \text{ J/K/kg}$$

$$2^{\text{nd}} \text{ term is } -0.32 \text{ J/K/kg}$$

$$\Delta S = 12.5 \text{ J/K/kg}$$